

## Highly Oxygenated Monoterpenes from *Eupatorium fortunei*

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**Abstract:** A pair of epimers of highly-oxygenated monoterpenes were isolated from the traditional Chinese medicine *Eupatorium fortunei*. Their structures were elucidated on the basis of the spectral analysis as (1R\*, 2S\*, 3R\*, 4R\*, 6S\*)-1, 2, 3, 6-tetrahydroxy-*p*-menthane (**1**) and (1S\*, 2S\*, 3S\*, 4R\*, 6R\*)-1, 2, 3, 6-tetrahydroxy-*p*-menthane (**2**).

**Keywords:** *Eupatorium fortunei*, traditional Chinese medicine, epimer, highly-oxygenated monoterpenes.

*Eupatorium fortunei* is a traditional Chinese medicine, which is widespread in China and used for the treatment of dropsical swelling, chills and fever, as a diuretic, antipyretic<sup>1</sup>. A Japanese research group has found a series of thymol derivatives from *Eupatorium fortunei*<sup>2</sup>. In order to search for its biologically active constituents our group carried out a further phytochemical study on this plant and two new highly-oxygenated monoterpenes were isolated from the EtOAc part of its MeOH extract. The EtOAc part showed weak cytotoxicity: IC<sub>50</sub>=69.5 and 86.1 μg/mL against HL-60 and SMMC-7721 cells by SRB method.

Compound **1** was obtained as colorless oil,  $[\alpha]_D^{14} +4$  (*c* 0.15, CH<sub>2</sub>Cl<sub>2</sub>). Its IR spectrum showed obvious absorption band for hydroxy groups at 3423 cm<sup>-1</sup>. The HRSIMS showed  $[M+H]^+$  at *m/z* 205.0536 (calcd. for C<sub>10</sub>H<sub>20</sub>O<sub>4</sub>+H<sup>+</sup> 205.1434) and the EIMS gave a molecular ion peak at *m/z* 204 and fragment ion peaks at *m/z* 186  $[M-H_2O]^+$ , 168  $[M-2H_2O]^+$ , 153  $[M-2H_2O-Me]^+$ , 125  $[M-2H_2O-isopropyl]^+$  and 107  $[M-3H_2O-isopropyl]^+$ , corresponding to a molecular formula C<sub>10</sub>H<sub>20</sub>O<sub>4</sub>. The <sup>1</sup>H-NMR, <sup>13</sup>C-NMR and DEPT spectra (**Table 1**) showed signals for a quaternary carbon, five methines, a methylene and three methyls. The proton signals of three methyl groups were at δ<sub>H</sub> 1.39 (s, 3H), 0.92 (d, 3H, *J* = 7.2 Hz), 0.78 (d, 3H, *J* = 7.2 Hz) and three of five methine protons were bearing oxygen signals at δ<sub>H</sub> 3.69 (d, 1H, *J* = 9.3 Hz), 3.96 (dd, 1H, *J* = 9.3, 11.4 Hz) and 3.77 (t, 1H, *J* = 2.4 Hz) in the <sup>1</sup>H-NMR spectrum. The chemical shift of the quaternary carbon was at δ<sub>C</sub> 74.6 in the <sup>13</sup>C-NMR spectrum. Thus compound **1** was deduced as a menthane monoterpene derivative with four hydroxys<sup>3</sup>. The cross-peaks of H-3 (δ 3.96) with H-2 (δ 3.69) and H-4 (δ 1.97); H-4 with H-3, H-5 (δ 1.59, 1.77) and

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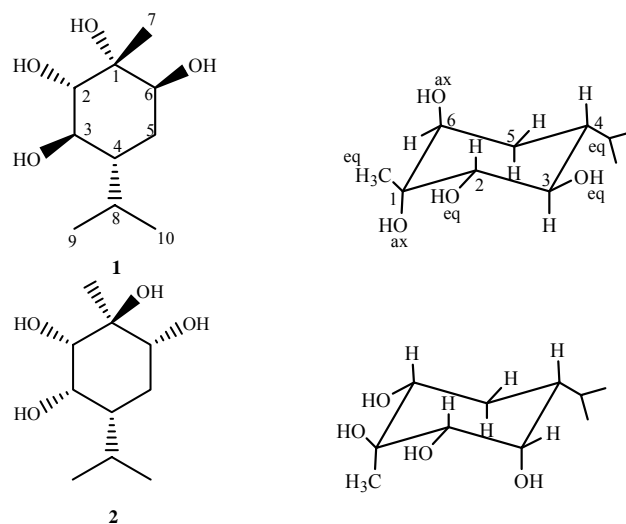
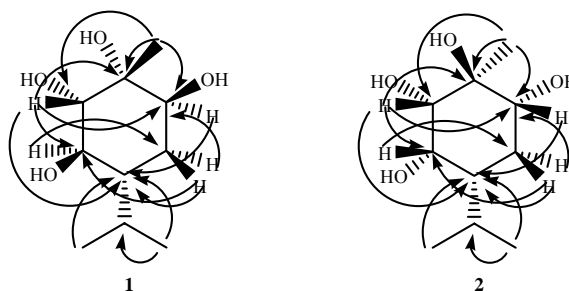
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H-8 ( $\delta$  2.30); H-6 ( $\delta$  3.77) with H-5 in the  $^1\text{H}$ - $^1\text{H}$  COSY of **1** and the correlations of a methyl singlet at  $\delta_{\text{H}}$  1.39 with C-1 ( $\delta$  74.6), C-2 ( $\delta$  76.7) and C-6 ( $\delta$  72.9) as well as other correlations in the HMBC (**Figure 2**) showed that the four hydroxys were located at C-1, C-2, C-3 and C-6, respectively and the isopropyl group was present on C-4, indicated that compound **1** was 1, 2, 3, 6-tetrahydroxy-*p*-menthane. The relative stereo-structure of **1** could be determined by the coupling pattern of cyclic protons in its  $^1\text{H}$ -NMR spectrum. The large coupling constants of H-3 with H-2 and H-4 ( $J_{3,2} = 9.3$  Hz,  $J_{3,4} = 11.4$  Hz) showed that C<sub>2</sub>-H, C<sub>3</sub>-H and C<sub>4</sub>-H were axial protons, and the small coupling constants of H-6 with H-5<sub>ax</sub> and H-5<sub>eq</sub> ( $J_{6,5a} = J_{6,5e} = 2.4$  Hz) was characteristic of the equatorial H-6 (**Figure 1**). The configuration of C<sub>1</sub> was determined from NOE experiments: irradiation of the methyl group at  $\delta$  1.39 (H-7) enhanced the signal at  $\delta$  3.69 (H-2). Therefore the structure of **1** was identified as (1R\*, 2S\*, 3R\*, 4R\*, 6S\*)-1, 2, 3, 6-tetrahydroxy-*p*-menthane.

Compound **2** was obtained as colorless oil,  $[\alpha]_{\text{D}}^{17} -36$  (*c* 1.4, CH<sub>2</sub>Cl<sub>2</sub>). Its IR and MS were very similar to these of **1**. The molecular formula was the same as that of compound **1** (C<sub>10</sub>H<sub>20</sub>O<sub>4</sub>). Its HRSIMS due to  $[\text{M}+\text{H}]^+$  was at  $m/z$  205.0521 (calcd. for  $[\text{C}_{10}\text{H}_{20}\text{O}_4+\text{H}]^+$  205.1434) and EIMS due to  $[\text{M}]^+$  was at  $m/z$  204 and fragment ion peaks at  $m/z$  189  $[\text{M}-\text{Me}]^+$ , 186  $[\text{M}-\text{H}_2\text{O}]^+$ , 161  $[\text{M}-\text{isopropyl}]^+$ , 143  $[\text{M}-\text{H}_2\text{O}-\text{isopropyl}]^+$  and 125  $[\text{M}-2\text{H}_2\text{O}-\text{isopropyl}]^+$ . **2** and **1** had similar chemical shifts of protons but obvious different coupling constants in  $^1\text{H}$ -NMR (**Table 1**), and their shifts of carbons had small difference in  $^{13}\text{C}$ -NMR (**Table 1**), which indicated that compound **2** was a stereo-isomer of **1**. The protons and carbons were easily assigned by HMBC (**Figure 2**) and  $^1\text{H}$ - $^1\text{H}$  COSY experiments. Stereochemically, the coupling constants of H-4 with H-5<sub>ax</sub> ( $J_{4,5a} = 13.5$  Hz), H-6 with H-5<sub>ax</sub> ( $J_{6,5a} = 10.8$  Hz) and H-3 with H-2 and H-4 ( $J_{3,2} = J_{3,4} = 2.7$  Hz) showed that C<sub>4</sub>-H and C<sub>6</sub>-H were axial protons and C<sub>3</sub>-H was equatorial proton. In the NOE experiment, the enhancement between H-6 and H-2 could be observed but no enhancement between the methyl at  $\delta$  1.45 (H-7) and H-6 or H-2, indicating that the methyl (H-7) was at opposite side of C<sub>2</sub>-H and C<sub>6</sub>-H. Thus compound **2** was deduced to be (1S\*, 2S\*, 3S\*, 4R\*, 6R\*)-1, 2, 3, 6-tetrahydroxy-*p*-menthane.

**Table 1**  $^1\text{H}$ -NMR (300 MHz),  $^{13}\text{C}$ -NMR (100 MHz) and DEPT data of compounds **1** and **2** (CDCl<sub>3</sub>,  $\delta$  ppm, J Hz)

No.	<b>1</b> $\delta_{\text{H}}$	<b>2</b> $\delta_{\text{H}}$	No.	<b>1</b> $\delta_{\text{C}}$	<b>2</b> $\delta_{\text{C}}$	DEPT
1			1	74.6	74.3	C
2	3.69 d (9.3)	3.81 d (2.7)	2	76.7	77.8	CH
3	3.96 dd (9.3, 11.4)	4.15 t (2.7)	3	68.9	70.4	CH
4	1.97 ddt (11.7, 11.4, 3)	2.06 dt (13.5, 2.7)	4	41.4	43.5	CH
5a	1.77 m	1.70 m	5	26.8	32.7	CH <sub>2</sub>
5e	1.59 m	1.78 m				
6	3.77 t (2.4)	4.29 dd (10.8, 2.4)	6	72.9	66.9	CH
7	1.39 s	1.45 s	7	23.9	20.6	CH <sub>3</sub>
8	2.30 m	1.68 m	8	27.8	27.8	CH
9	0.92 d (7.2)	0.97 d (6.6)	9	20.9	20.9	CH <sub>3</sub>
10	0.78 d (7.2)	0.97 d (6.6)	10	14.9	20.9	CH <sub>3</sub>

**Figure 1** Structures of compounds **1** and **2****Figure 2** Correlation of HMBC of compounds **1** and **2****Acknowledgment**

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